THIOPAQ O&G



Biodesulfurization system

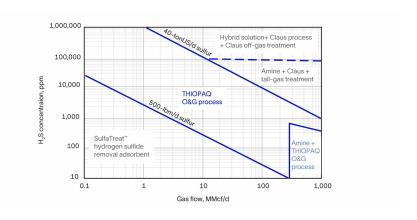
Applications

- → Natural gas pressure range between 2-psi and 1,300-psi [0.01-MPa and 9-MPa] gauge pressure
- → Debottlenecking of sulfur recovery units
- → Treatment of Claus process tail gas
- → Replacement or conversion of alternative aqueous processes that are prone to plugging

Features

The process removes $\rm H_2\,S$ from low-, medium-, and high-pressure streams in direct treat operations. It also easily installs for indirect treating and runs downstream of amine, Claus, and membrane units for emissions cleanup and sulfur recovery. The THIOPAQ O&G system produces treated gas that meets a typical $\rm H_2S$ outlet specification of 4 ppm or less.

The process liquid that is home to the bacteria is regulated by a programmable logic controller (PLC), which monitors pH, conductivity, temperature, oxygen demand, and many other parameters. Should any of these parameters fall outside its operational range because of changes in gas flow rates, total sulfur loading, or both, the PLC automatically adjusts the parameter to ensure that the bacteria continue their highly efficient conversion cycles.



H₂S removal technology selection.

How it improves performance

THIOPAQ $^{\circledR}$ O&G* biodesulfurization system uses nonhazardous, naturally occurring, self-regulating bacteria to sweeten gas by converting H2S in a produced gas stream to manageable solid elemental sulfur. The process also includes extraction of the sulfur from the system, for use in a variety of agricultural applications or disposal in a landfill. It thus addresses two goals: environmental stewardship and positive economics.

The process system is highly efficient in converting hydrogen sulfide to sulfur. Compared with typical caustic scrubber systems, only modest amounts of caustic addition are required to maintain system alkalinity and pH for absorption. No heating equipment is used for regeneration and no catalyst is needed for conversions. This approach successfully reduces system cost. Small amounts of nutrients are added to the system to keep the bacteria in good health. After the one-time addition of bacteria to the system, the colony expands or contracts based on the feeding of $\rm H_2S$ into the system. This enables the biological system to be self-regulating, supporting significant process turndown if conditions demand.

Reliability

- → Process requires less equipment compared with conventional desulfurization processes.
- → Biological sulfur's hydrophilic nature eliminates plugging or fouling issues.
- → Bacteria are naturally occurring, robust, self sustaining, and self regulating.

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Simplicity of operation

- → Nontoxic process provides you with simple operation and low maintenance.
- → Operating parameters are easy to control.
- → Supervision requirements are minimal.
- → Massive buffering capacity minimizes the impact of upsets.
- → Wide turndown in gas flow and H₂S inlet concentration can be easily accommodated.

Low operating costs

- → Chemical requirement is much lower compared with alternative aqueous technologies.
- → There is less equipment to maintain and operate compared with conventional amine or Claus technology.

Simplicity of design

- → Process operates across a wide range of pressures and does not require compression.
- → Gas purification and sulfur recovery are integrated in one process.
- → Filtering carbon or particulates is not required.
- → Process regeneration does not require heat.

Enhanced safety

- → Nontoxic wastestreams are disposed of more easily compared with conventional sweetening processes.
- → H₂S is not concentrated at any time during the process.
- → H₂S is physically bound to the gas scrubbing solution.

Environmental consciousness

- → Naturally occurring biological process is an environmentally friendly green technology.
- → Air vent gas contains less than 1-ppm H₂S by volume.
- → Process is ideally suited to environmentally sensitive areas where venting, incineration, or reinjection of the H₂S are not desirable.
- → Biological sulfur by-product can be used for agricultural purposes, either in the form of a slurry or low-purity sulfur cake.
- → Environmental Technology Verification (ETV)—produced in association with the US Environmental Protection Agency (EPA)—enables easier regulatory compliance and approval.

How it works

H₂S absorption by aqueous soda solution

To begin, a pH-controlled aqueous soda solution (THIOPAQ solution) flows into a contactor containing plastic packing. Within the vessel, sour gas flows in a countercurrent direction to the THIOPAQ solution. The aqueous solution absorbs

H₂S from the natural gas stream, forming sulfide ions (HS⁻), and the sweetened gas

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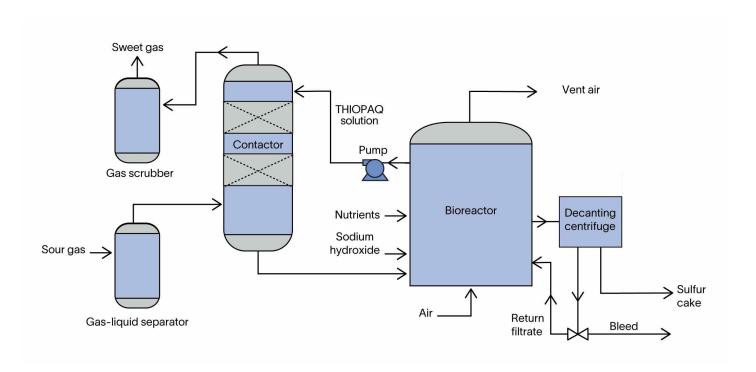
stream exits the contactor with as much as 99% of the H_2S removed. The contactor is the only place where H_2S is present in the process other than in some gas inlet scrubbing vessels that are required upstream to better clean the gas for improved H_2S removal.

Conversion of hydrogen sulfide ions into elemental sulfur

The second part of the process is conversion of the hydrogen sulfide ions that formed in the process solution. This conversion occurs when the solution is pumped into bioreactors where the haloalkaliphilic strains of the Thioalkalivibrio bacteria are stimulated by oxygen delivered by variable-speed air blowers. The bacteria oxidize and covert the sulfide ions into elemental sulfur. An enzyme is produced during the biological conversion that covers the sulfur particles, making them nonsticky or hydrophilic. This is unlike other liquid processes, which require chemicals for conversion and produce sticky or hydrophobic sulfur. Because sulfur particles are present throughout the process wherever the liquid solution exists, it follows that nonsticky sulfur particles lead to significantly less downtime for plant maintenance.

Removal of sulfur particles

The final step is removal of the converted sulfur particles. Testing and experience have shown that compared with vacuum filters and filter presses, the decanting centrifuge is the most efficient and cost-effective method of removing sulfur particles. The supernatant liquid is returned to the system to minimize requirements for additional fluids.



THIOPAQ O&G biodesulfurization process flow diagram.